

SYNTHESIS OF RING SYSTEM RELATED TO HASUBANAN ALKALOID

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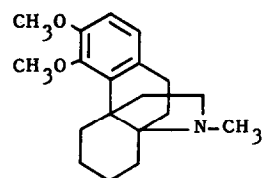
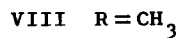
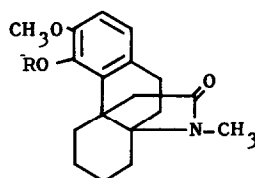
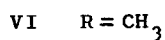
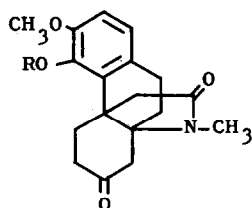
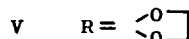
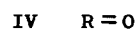
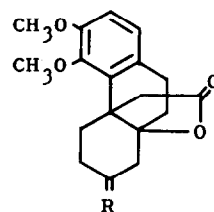
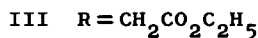
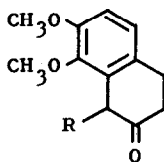
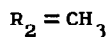
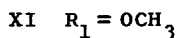
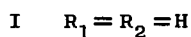
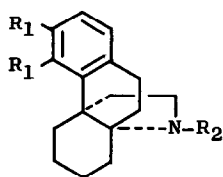
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Alkaloids containing hasubanan framework (I) were isolated from various *Stephania* species and the complete constitutions were settled recently<sup>1)</sup>. In this paper the authors wish to communicate the synthesis of dl-N-methyl-hasubanan derivative<sup>\*1</sup>.

Condensation of pyrrolidine enamine of 7,8-dimethoxy-2-tetralone (II)<sup>2)</sup> with ethyl bromoacetate, followed by hydrolysis, afforded the keto ester (III), m.p. 68-68.5°, C<sub>16</sub>H<sub>20</sub>O<sub>5</sub>. By treatment with 1-diethylamino-3-butanone methiodide and potassium ethoxide<sup>3)</sup>, the keto ester (III) was converted into the keto lactone (IV), m.p. 160-163°, IR  $\nu_{\max}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 1767 (lactone); 1722 (ketone), C<sub>18</sub>H<sub>20</sub>O<sub>5</sub>. Ketalization of the keto lactone (IV) by ethylene glycol in the presence of p-toluenesulfonic acid in benzene gave the ketal lactone (V), m.p. 194-196°, C<sub>20</sub>H<sub>24</sub>O<sub>6</sub>, which on treatment with the mixture of methylamine and methylamine hydrochloride in aq. dioxane<sup>\*2</sup> gave the keto lactam (VI), m.p. 170-172°, IR  $\nu_{\max}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 1718 (ketone); 1675 (lactam), NMR (CDCl<sub>3</sub>)  $\tau$ : 7.20 (N-CH<sub>3</sub>), C<sub>19</sub>H<sub>23</sub>O<sub>4</sub>N, along with the O-demethylated keto lactam (VII)<sup>\*3</sup>, m.p. 247-250°, IR  $\nu_{\max}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 3500 (hydroxyl); 1715 (ketone); 1675 (lactam), C<sub>18</sub>H<sub>21</sub>O<sub>4</sub>N.

Huang Minlon reduction of the keto lactam (VI) afforded the lactam (VIII), m.p. 187.5-190°, C<sub>19</sub>H<sub>25</sub>O<sub>3</sub>N, and the O-demethylated product (IX)<sup>\*3</sup>, m.p. 167-170°, C<sub>18</sub>H<sub>23</sub>O<sub>3</sub>N. Methylation of IX with methyl iodide and potassium carbonate



in acetone gave VIII. Treatment of the lactam (VIII) with lithium aluminum hydride gave the dl-3,4-dimethoxy-N-methylhasubanan (X) as a colorless oily substance, which was characterized as its hydrobromide, m.p. 242-250°,  $C_{19}H_{27}O_2N.HBr$ . The over-all yield of X from II was approximately 2.2%.

The racemic product (X) and the 3,4-dimethoxy-N-methylhasubanan (XI)<sup>4)</sup> derived from naturally occurring hasubanan alkaloids were found to be identical in terms of their IR spectra (in  $CHCl_3$ ), NMR spectra (in  $CDCl_3$ ), and the thin layer chromatographic behaviors.

The synthesis of dl-hasubanan derivative by the above route points to the accessibility of similar racemic hasubanan derivatives.

- \*1 All compounds given by formulas in this communication gave satisfactory elementary analyses.
- \*2 The expectation that the keto lactone (IV) could be transformed into the keto lactam (VI) was not fulfilled, and the experimental condition will be presented in the full paper.
- \*3 M. Gates, et al. (J. Am. Chem. Soc., 78, 1380 (1956).) and M. Tomita et al.<sup>4)</sup> have reported that the methoxyl group at C-4 position in morphinan and hasubanan series alkaloids was demethylated under alkaline condition. Furthermore, VII and IX showed a strong blue color with 2,6-dichloroquinone-4-chloroimide suggesting hydroxyl groups should be situated at C-4 position (cf. homostephanoline<sup>1a)</sup> which has hydroxyl group at C-3 position was negative to the reagent).

#### REFERENCES

- 1) a) T. Ibuka and M. Kitano, Chem. Pharm. Bull. (Tokyo), 15, 1809, 1937, 1944 (1967), and the references cited therein.  
b) H. L. de Waal, B. J. Prinsloo and R. R. Arndt, Tetrahedron Letters, 6169 (1966).  
c) D. H. R. Barton, G. W. Kirby and A. Wiechers, J. Chem. Soc. (C), 1966, 2313.  
d) M. Tomita and M. Kozuka, Tetrahedron Letters, 6229 (1966); Idem., Yakugaku Zasshi, 87, 1203 (1967).
- 2) M. D. Soffer, R. A. Stewart, J. C. Cavagnol, H. G. Gellerson and E. A. Bowler, J. Am. Chem. Soc. 72, 3704 (1950).
- 3) J. W. Cornforth and R. Robinson, J. Chem. Soc. 1949, 1855.
- 4) M. Tomita, T. Ibuka, Y. Inubushi, Y. Watanabe and M. Matsui, Chem. Pharm. Bull. (Tokyo), 13, 538 (1965); M. Tomita, T. Ibuka, Y. Inubushi and K. Takeda, Chem. Pharm. Bull. (Tokyo), 13, 695, 704 (1965).